

# **Product Information Sheet**

# T8224 Tris-Acetate-Phosphate

Synonym: TAP

#### **Properties:**

Form:	Fine to Coarse Powder
Appearance:	White Powder
Application:	Freshwater algal culture
Solubility:	Few to no insolubles at the typical working concentration in water
Typical Working	3.17 g/L
Concentration:	5.17 y/L
Storage Temp:	2 - 8°C
Storage Temp of	Preparation of concentrated solutions is not recommended as insoluble precipitates may
Stock Solution:	form.
Biological Assay:	Algal culture tested with Chlamydomonas reinhardtii

#### Formula (mg/L):

400.0	EDTA, Disodium Salt	50.00
2420	Ferrous Sulfate•7H2O	4.990
1.100	Magnesium Sulfate, Anhydrous	48.83
11.40	Manganese Chloride•4H2O	5.060
37.74	Potassium Phosphate, Dibasic	108.0
1.610	Potassium Phosphate, Monobasic	54.00
1.570	Zinc Sulfate•7H2O	22.00
	2420 1.100 11.40 37.74 1.610	2420Ferrous Sulfate•7H2O1.100Magnesium Sulfate, Anhydrous11.40Manganese Chloride•4H2O37.74Potassium Phosphate, Dibasic1.610Potassium Phosphate, Monobasic

## **Application Notes:**

Tris-acetate-phosphate medium (TAP) is a standard maintenance medium often used for Chlamydomonas reinhardtii, the most well-characterized eukaryotic freshwater algae. Ammonium (NH<sub>4</sub><sup>+</sup>) serves as the primary nitrogen source and Tris buffers the pH. Since TAP contains a relatively low concentration of phosphate, it can be used for <sup>32</sup>P labeling as well as experiments/isolations that require clarity of solid-substrate cultures (e.g. agar) (Harris 1989).

### Media Preparation:

The standard (photoheterotrophic) medium is prepared as follows: 1 mL of glacial acetic acid (A256) is added per liter of medium to obtain the proper final concentration of acetate (17.4 mM). The final solution pH is adjusted to 7.0 +/- 0.1 with HCl.

If T8224 is to be used photoautotrophically, glacial acetic acid is omitted. The final solution pH is adjusted to 7.0 +/- 0.1 with HCl.

PhytoTech Labs Inc. also carries TAP medium in liquid form, Product No. T8050.

## **References:**

Gorman, D.S., and R.P. Levine (1965) Proc. Natl. Acad. Sci. USA 54, 1665-1669. Harris, E.H. (1989): The *Chlamydomonas* sourcebook: a comprehensive guide to biology and laboratory use. Academic Press, San Diego, 780pp.

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